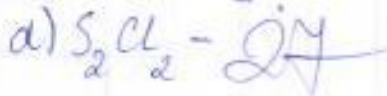
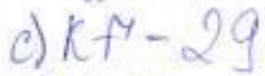
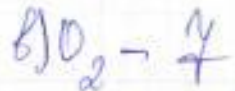
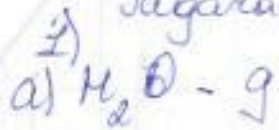
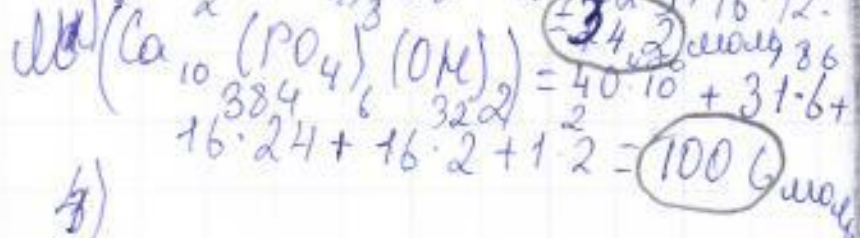
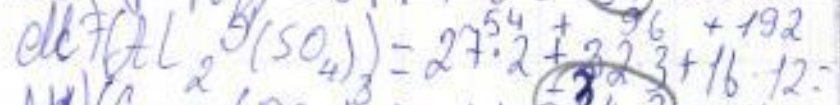
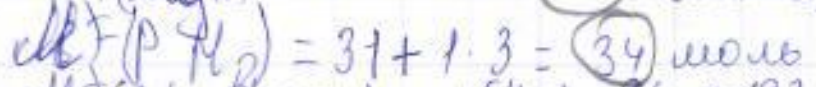
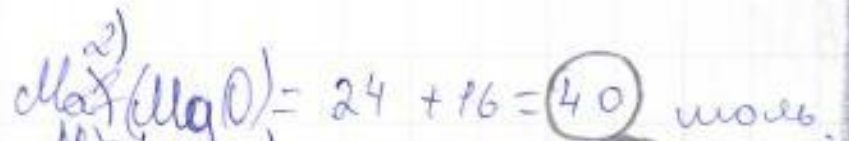
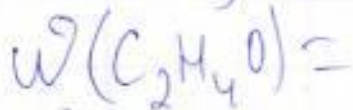
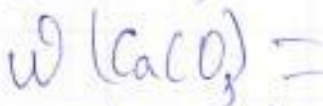
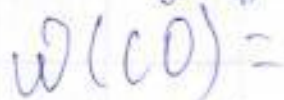
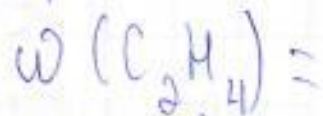


Задача 1



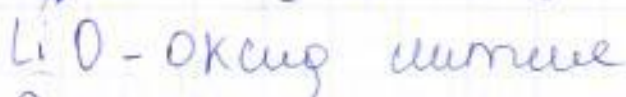
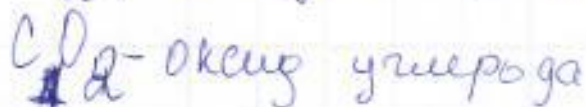
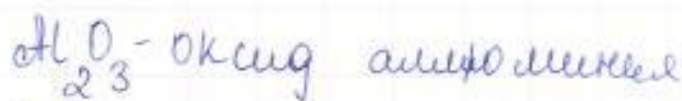
3)



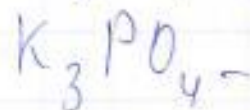
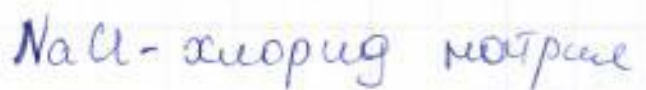
4)

 S_8 - простое вещество Al_2O_3 - простое вещество P_4 - простое вещество Ne - простое вещество

Задача 2



2)



Задача 1

(18)

- a) $H_2O - 9$
- b) $O - 7$
- c) $K^+ - 29$
- d) $S_2Cl_2 - 27$

- 2)
- дл $(H_2O) = 2 \cdot 1 + 16 = 18$
 - дл $(PH_3) = 31 + 1 \cdot 3 = 34$ моле
 - дл $(Al_2SO_4)_3 = 27 \cdot 2 + 32 \cdot 3 + 16 \cdot 12 = 342$ моле
 - дл $(Ca_{10}(PO_4)_6(OH)_2) = 40 \cdot 10 + 31 \cdot 6 + 16 \cdot 24 + 16 \cdot 2 + 1 \cdot 2 = 1006$ моле

3)

- ω(C_2H_4)
- ω(CO)
- ω($CaCO_3$)
- ω(C_2H_4O)

4)

- S_8 - простое вещество
- NO_2 - простое вещество
- P_4 - простое вещество
- He - простое вещество

Задача 2

- 1) Оксид марганца - Mn_2O_3
- 2) Оксид алюминия - Al_2O_3
- 3) Оксид углерода - CO_2
- 4) Оксид лития - Li_2O
- 5) Оксид фосфора - P_2O_5

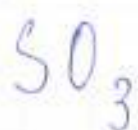
2)

- $NaCl$ - хлорид натрия
- $MgSO_4$ - сульфат магния
- K_3PO_4
- $CaCO_3$
- Al_2S_3

Задача 3

1)

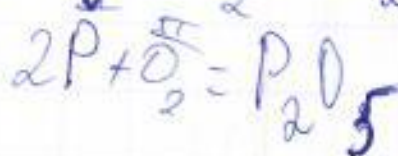
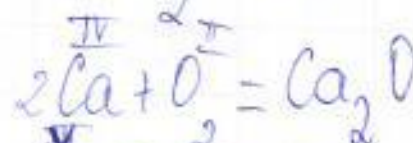
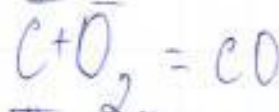
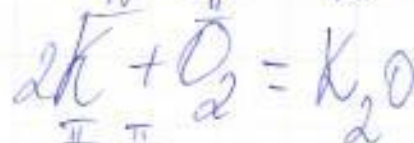
Оснoвные



Жеңiмдiктер



2)



Задача 4

Задача 3 (18)

1) Основные | Кислотные

